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Original article

Design and evaluation of herbal hepatoprotective formulation against paracetamol induced liver toxicity

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ABSTRACT

Aim: To isolate and identify the quercetin from polyherbal hepatoprotective formulation. Polyherbal formulations were developed by using five bioactive fractionated extracts of *Butea monosperma*, *Bauhinia variegata* and *Ocimum gratissimum* for treatment of liver disorders by exploiting the knowledge of traditional system of medicine and evaluated for hepatoprotective activity using acute liver toxicity model of paracetamol induced liver damage in rats.

Methods: Major active fractions were isolated by solvent fractionation and quantified by HPTLC method. Two polyherbal tablet formulations were developed by the wet granulation method using microcrystalline cellulose, aerosil and other excipients and subjected for physicochemical evaluation to assess physical stability followed by pharmacological screening. The prepared tablets were finally subjected to stability testing to assess its shelf-life. The rats were monitored for change in liver morphology, biochemical parameters like serum glutamate pyruvate transaminase (SGPT), serum glutamate oxaloacetate transaminase (SGOT), alkaline phosphatase (ALP) and total bilirubin for polyherbal tablet formulation at 50 mg/kg and polyherbal tablet formulation at 100 mg/kg.

Results: Active principle was isolated, quantified by HPTLC and characterized with IR. Both formulations showed significant hepatoprotective activity. The histological studies were also support the biochemical parameters. From the results of biochemical analysis and histopathological studies, it can be accomplished that polyherbal tablet formulation at 100 mg/kg can be effectively formulated into a suitable dosage form with added benefit of no side effects for control and cure of chronic ailments like liver disorders. A comparative histopathological study of liver exhibited almost normal architecture as compared to toxicant group.

Conclusion: Biochemical marker showed improved results for polyherbal tablet formulation at 100 mg/kg. Polyherbal tablet formulation contains a potent hepatoprotective agent suggested to be a flavone concentrated in polyherbal formulation which may find clinical application in amelioration of paracetamol induced liver damage.

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1. Introduction

In the last few decades there has been an exponential growth in the field of herbal medicine. It is getting popularized in developing and developed countries owing to its natural origin and lesser side effects.¹ Herbal remedies provide rational means for the treatment of many internal diseases which are considered to be stubborn and incurable in other system of medicines. It lays a great deal of emphasis upon the maintenance of positive health of an individual. It thus aims at both the prevention and cure of diseases.² An ancient system of traditional medicines like Siddha, Ayurveda, Chinese and Japanese have been adopted for the diagnosis, prevention and treatment of liver disorders. This attempts to prove scientific insight behind the traditional adaption. Less toxicity, better therapeutic effect, good patient compliance and cost effectiveness are the reasons for choosing drug from natural origin.³

Ayurvedic and herbal medicinal products contain a combination of number of chemical compounds that may give the anticipated activity in combination.⁴

The development of a stable polyherbal formulation is a challenging task because of the large number of varied chemical

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compounds present in the different medicinal plants. Hence, the entire herbal drug or herbal drug preparation is regarded as active drug substance, regardless of whether constituents with defined therapeutic activity are known and unfortunately, the quality of a majority of them remains uncontrolled. All these issues have been acknowledged in the draft of the Strategic Plan for Regional Traditional Medicine of the World Health Organization.⁵

A great deal of research has been carried out to evaluate scientific basis for the claimed hepatoprotective activity of herbal agents as in the form of polyherbal formulation. *Butea monosperma* (Fabaceae) is commonly known as *palas* and is beneficial in conditions of liver disease and inflammation.⁶ The plant mainly contains flavones (quercetin),⁷ kino-tannic acid, and gallic acid.⁸

Bauhinia variegata (Leguminosae) commonly known as Kachnar, is found to be beneficial in Ayurveda as tonic to the liver and antiinflammatory, healing activity, antioxidant activity.⁹ *B. variegata* has been reported to contain quercetin, rutin, apigenin and apigenin 7-O-glucoside. Flavonoids and quercetin in particular are potent antioxidants and are known to modulate the activities of various enzyme systems due to their interaction with various biomolecules.¹⁰

Ocimum gratissimum Linn. (Lamiaceae) is known traditionally for its effect in liver diseases, inflammation and antibacterial activity.¹¹ The active constituents, flavonoids, quercetin 3-O-glucoside, rutin, kaempferol 3-O-rutinoside and vicenin-2 were identified.¹²

Quercetin (flavonoid) and ursolic acid (triterpenic acid), well known for its hepatoprotective effects in both acute chemically induced liver injury and chronic liver fibrosis and cirrhosis.¹³

Ursolic acid was previously determined in dichloromethane and ethyl acetate fractions of methanolic extract of *O. gratissimum* and in developed herbal hepatoprotective tablet in my previous article.⁸

The present study was designed to develop polyherbal formulation taking advantage of ayurvedic principles, where herbal ingredients present in plants are known to have specific activity in modulation of liver disease condition so the present study was undertaken to evaluate the hepatoprotective effect of developed polyherbal formulation in which acute hepatotoxicity was induced by paracetamol treatment.

2. Material and methods

2.1. Material

The fresh bark of *B. monosperma*, *B. variegata* and fresh leaves of *O. gratissimum* were collected from Maliba Pharmacy College Campus. Voucher specimen (No: MPC/13032010/01, 02 and 03) has been deposited in the Department of Bioscience, Veer Narmad South Gujarat University, Surat, India. Rats were used for hepatoprotective study, with prior approval from the Institutional Animal Ethical Committee (Registration No. 717/02/a/CPCSEA/30 Jan 2010) of Maliba Pharmacy College, Uka Tarsadia University.

2.2. Extraction

All the three crude drugs were extracted with alcohol and then alcoholic extract of each plant was subjected to solvent fractionation.

B. monosperma: Ethanolic extract obtained was fractionated with petroleum ether, benzene, chloroform and acetone (AcO) in the order of increasing polarity to obtain respective fractions.¹⁴

B. variegata: Methanolic extract was fractionated with hexane, ethyl acetate (EtOAc) and n-butanol (n-ButOH) in the order of their increasing polarity to obtain respective fractions.¹⁵

O. gratissimum: Methanolic extract of *O.* gratissimum was fractionated with hexane, dichloromethane(DCM), ethyl acetate (EtOAc) and methanol in the order of their increasing polarity to obtain respective fractions.¹⁶

2.3. Establishment of qualitative phytoprofile of fractionated extracts

2.3.1. Qualitative phytochemical analysis

Each fraction was subjected to various qualitative chemical tests using reported methods to determine the presence or absence of metabolites viz., alkaloids, tannins, flavonoid, steroid, terpenoids and phenolic compounds etc.¹⁷

2.3.1.1. Chemical test for flavonoids in each fraction

- Dilute ammonia solution 5 mL was added to a portion of each extract fraction followed by addition of concentrated H₂SO₄ a yellow coloration observed indicated the presence of flavonoids.
- ii. The yellow coloration disappeared on standing. Few drops of 1% aluminum solution were added to a portion of each fraction. A yellow coloration was observed indicating the presence of flavonoid.¹⁸

2.4. Qualitative phytochemical analysis

2.4.1. Determination of total phenols

Acetone fraction of *B. monosperma*, ethyl acetate and n-butanol fractions of *B. variegata* and dichloromethane and ethyl acetate fractions of *O. gratissimum* (100 μ L) of each fraction were mixed with 1 mL Folin–Ciocalteu reagent and 0.8 mL of 7.5% Na₂CO₃. The resultant mixture of each fraction was measured at 765 nm after 2 h at room temperature. The mean of three readings was used and the total phenolic content was expressed in milligram of gallic acid equivalents/1 g extract. The coefficient of determination was found to be $r^2 = 0.992.^{19}$

2.4.2. Determination of total flavonoids

Standard quercetin was used to make the calibration curve [0.04, 0.02, 0.0025 and 0.00125 mg/mL in 80% ethanol (v/v)]. The standard solutions and acetone fraction of *B. monosperma*, ethyl acetate and n-butanol fractions of *B. variegata* and dichloromethane and ethyl acetate fractions of *O. gratissimum* (0.5 mL) of each fraction was mixed with 1.5 mL of 95% ethanol (v/v), 0.1 mL of 10% aluminum chloride (w/v), 0.1 mL of 1 mol/L sodium acetate and 2.8 mL water. The volume of 10% aluminum chloride was substituted by the same volume of distilled water in blank. After incubation at room temperature for 30 min, the absorbance of the reaction mixture of each fraction and standard solution was measured at 415 nm. The mean of three readings was used and the total flavonoid content was expressed in milligram of quercetin equivalents/1 g extract. The coefficient of determination was $r^2 = 0.990.^{20}$

2.5. Preparation of formulations

Polyherbal tablet formulations (PTF) contain the crude raw materials of *B. monosperma*, *B. variegata* and *O. gratissimum* prepared by wet granulation method using suitable excipients like microcrystalline cellulose, starch, crospovidone, aerosil and magnesium stearate.²¹ The composition of tablet formulations is shown in Table 1.

2.6. Quantification of quercetin by HPTLC

A Camag HPTLC system was used with beam size of 0.4×0.4 mm, and wavelength of 380 nm. Silica gel 60 F 254 $(20 \times 20 \text{ cm}, 0.25\text{-mm} \text{ thick}, \text{Merck})$ plates were used for HPTLC. Samples were applied (15 mm from the bottom edge of the plate) in the form of 6 mm bandwidth with Hamilton svringes using Linomat V applicator. A constant application rate of 150 nl/s was supplied and a space of 10 mm was kept between two bands. The mobile phase constituted of toluene:ethyl acetate:formic acid, 5:4:0.1 (v/v/v). Linear ascending development was carried out in a twin trough glass chamber saturated with mobile phase. The optimized saturation time for mobile phase was 30 min at room temperature (25 °C \pm 2 °C) with relative humidity of 55 \pm 5%. After developing TLC plate a current of air was used to dry it. Densitometric scanning was performed on Camag TLC scanner III in absorbance mode at 380 nm. The source of light was deuterium beam. The quantitative determination was performed by using the external calibration method. A stock solution of quercetin (500 µg/ mL) was prepared in methanol. Different volume of stock solution i.e. 0.5, 1.0, 1.5, 2.0, 2.5 in µg/spot were spotted on TLC plate of 200, 400, 800, 1600 and 3200 ng/spot of quercetin. Calibration curve range $(0.5-2.5 \mu g/spot)$ for quercetin was found to be linear. The data of peak areas plotted against the corresponding concentrations were collected. To determine the content of quercetin in extract 10 mg of each fraction was transferred in 10 mL volumetric flask containing 8 mL methanol, vortexed for 10 min and diluted to 10 mL with methanol. The resulting solution was filtered through Whatman filter paper. The 15 µL of the filtered solution was applied on TLC plate subsequently developed and scanned as described above.

2.7. Evaluation of power blends

The polyherbal tablets were prepared by wet granulation method hence micromeritic studies (angle of repose, bulk density, tapped density, % compressibility) were evaluated prior to punching of tablets.

Table 1

Composition of polyherbal tablets formulations (PTF).

Sr. no	Ingredients	PTF-1	PTF-2
		Quantity (mg/tablet)	Quantity (mg/tablet)
	Dry mix		
1	Acetone fraction of	10	20
	Butea monosperma		
2	Ethyl acetate fraction	10	20
	of Bauhinia variegata		
3	n-Butanol fraction of	10	20
	Bauhinia variegata		
4	Dichloromethane of	10	20
_	Ocimum gratissimum		
5	Ethyl acetate fraction	10	20
<i>c</i>	of Ocimum gratissimum	005	075
6	Microcrystalline	325	275
7	cellulose Starch	50	50
8		50	50
δ	Crospovidone Granulation	20	20
9	Water	a .c	a c
9	Prelubrication	q.s.	q.s.
10	Starch	25	25
10	Aerosil	10	10
12	Talc	15	15
	Lubrication		
13	Magnesium stearate	5	5
Tablet we	eight (mg)	500	500

2.8. Physical characterization of polyherbal tablets

Polyherbal tablets systems were characterized for thickness, diameter, disintegration time, weight variation, and friability. Crushing strength, thickness and friability were determined by Monsanto hardness tester, digital vernier calipers and friabilator respectively.

2.9. Isolation and identification of quercetin from polyherbal tablet

Weight accurately 500 mg equivalent of polyherbal tablet was transferred to the 10 mL volumetric flask and dissolved in 10 mL methanol. This solution was sonicated for 10 min and filtered through Whatman No. 1 paper to get solution containing 10 mg/mL. Concentrated solution obtained from polyherbal tablet was subjected to preparative TLC. Preparative TLC were performed on 20 cm \times 10 cm TLC aluminum plate coated with 200 µm layer thickness of silica gel 60 F 254 using toluene:ethyl acetate:formic acid, 5:4:0.1 (v/v/v) as mobile phase. The quantification was carried out by densitometric scanning in absorbance mode at wavelength 380 nm. The silica in the respective marked area was scraped off and collected carefully in a test tube. The scraped silica, which contained the standard quercetin was extracted in methanol and filtered individually. The filtrates containing quercetin.

Melting point of quercetin was determined by open capillary method. Structural confirmation of the isolated quercetin was done by I.R. Spectroscopy and compared with that of standard.

2.10. In-vivo hepatoprotective activity of polyherbal tablet formulations against paracetamol induced toxicity

2.10.1. Test animals

Male Wistar rats (125–175 g) were used for determination of maximum tolerable dose (MTD) and evaluation of hepatoprotective activity. The animals were housed in polypropylene cages at 25 °C \pm 1 °C with the relative humidity of 55 \pm 5% under 12 h/12 h light/dark cycle. They were received a standard chow and water ad libitum during experimentation. The food was with-drawn on the day before the experiment, but free access of water was allowed. A minimum of six animals was used in each group. Throughout the experiments, animals were processed according to the suggested international ethical guidelines for the care of laboratory animals. The study protocol was approved by the Institutional Animal Ethics Committee according to the regulation of Committee for the Purpose of Control and Supervision of Experiments on Animals (MPC 1007: dated: 30/01/2010).

2.10.2. Preparation and administration of test samples

Approved IAEC method was adopted for screening of hepatoprotective activity. Dried fractions were suspended in 1% CMC solution before oral administration to animals. Animals were divided into five groups of six rats each were used for the study. Groups 1 and 2 received normal saline (1 mL/kg, orally) for 7 days. Groups 3, 4 and 5 received 100 mg/kg bw doses of silymarin, PTF1 (50 mg/kg bw doses) and PTF2 (100 mg/kg bw doses) orally, once a day for 7 days. On the fifth day, after the administration of the respective treatments, all the animals in groups II, III, IV and V were administered paracetamol (PCM) 2 g/kg orally. On the seventh day, the blood samples were collected via orbital sinus puncture for the estimation of biochemical marker enzymes. Then the liver was carefully isolated and cleaned off extraneous tissue and preserved in 10% neutral formalin and then subjected to histopathological studies.^{22,23}

2.10.3. Assessment of biochemical parameters of liver

The enzymatic parameters of serums like serum glutamate pyruvate transaminase (SGPT), serum glutamate oxaloacetate transaminase (SGOT) and serum alkaline phosphatase (ALP) and nonenzymatic parameter like total bilirubin (TB) were assayed according to standard methods.^{24,25}

2.10.4. Statistical analysis

The results are expressed as mean \pm S.E.M (N = 6). Statistical significance was determined by one-way analysis of variance with P < 0.01 and P < 0.05 considered significant followed by Dunnett Multiple Comparisons Test. The analysis was performed by Graph Pad InStat software.

2.10.5. Stability study

Stability studies were carried out at 25 °C \pm 2 °C/60% RH \pm 5% and 40 °C \pm 2 °C/75% RH \pm 5% for a specific time period of 30 days for selected (PF2) formulation. The selected samples were evaluated for physical appearance, hardness and disintegration time in every 10 days interval.

3. Results and discussion

3.1. Phytochemical screening

Preliminary phytochemical screening of alcoholic extract and its fractions showed the presence of flavonoids, steroids, terpenoids, tannins and phenolic compounds. The chemical tests and TLC analysis demonstrated that AcO fraction of B. monosperma, EtOAc fraction and n-ButOH fraction of B. variegata, DCM and EtOAc fractions of O. gratissimum were rich in phenolic compounds. The phenolic content in *B. monosperma* (acetone fraction), *B. variegata* (ethyl acetate and n-butanol fractions) and O. gratissimum (dichloromethane and ethyl acetate fractions) were found to be 452 \pm 1.6, 712.4 \pm 2.4, 442.5 \pm 1.1, 735 \pm 2.1 and 1365 \pm 1.4 mg gallic acid/1 g fraction respectively. The flavonoid content in B. monosperma (acetone fraction), B. variegata (ethyl acetate and nbutanol fractions) and O. gratissimum (dichloromethane and ethyl acetate fractions) were found to be 251 \pm 1.8, 417 \pm 2.2, 227 \pm 3.2, 394.5 ± 2.4 and 717 ± 5.2 mg quercetin/1 g fraction respectively. The phenol and flavonoid contents are responsible for hepatoprotective activity; hence these solvent fractions were selected for further study.

3.2. HPTLC analysis

In HPTLC chromatogram, all tracks for standard quercetin at wavelength 380 nm were shown in Fig. 1A. The Rf value of standard

quercetin was found to be 0.38 and peak area was 9726 (Fig. 1B). Acetone fraction of *B. monosperma* showed twelve peaks, the sixth peak Rf value (0.39) was coinciding with standard Rf value and its peak area was 9909 (Fig. 2A), ethyl acetate fraction of *B. variegata* showed eight peaks, the fourth peak Rf value (0.38) was coinciding with standard Rf value of guercetin and its peak area was 7898 (Fig. 2B), n-butanol fraction of *B. variegata* showed six peaks, the third peak Rf value (0.38) was coinciding with standard Rf value and its peak area was 7570 (Fig. 2C), dichloromethane fraction of O. gratissimum showed nine peaks, the third peak Rf value (0.38) was coinciding with standard Rf value and its peak area was 9246 (Fig. 2D), ethyl acetate fraction of O. gratissimum showed seven peaks, the third peak Rf value (0.39) was coinciding with standard Rf value and its peak area was 12427 (Fig. 2E). The concentrations of quercetin in acetone fraction of B. monosperma, ethyl acetate and nbutanol fractions of *B. variegata*, dichloromethane and ethyl acetate fractions of O. gratissimum were found to be 0.395, 0.1744, 0.1382, 0.3229, 0.6734 (µg/10 mg) respectively.

3.3. Pre- and post-compression evaluation parameters

Pre-compression parameters (angle of repose, bulk density, tapped density and % compressibility) of powder blends and tablets showed satisfactory micromeritics properties as angle of repose of was less than 30° and % compressibility was in the range of 15–17%. The physical properties (weight variation, thickness, diameter, crushing strength, friability) of polyherbal tablets shown in Table 2. Tablets are circular biconvex in shape, light greenish with few white spots and slightly dark greenish spots color for PTF-1 and light greenish with slightly dark greenish spots for PTF-2 tablets, odor and taste are characteristics for the formulations. They produced average weight between (506 ± 3.77 and 500.6 ± 0.02); uniform thickness, crushing strength and acceptable friability (less than 1%). Tablets were disintegrated between 2 and 4.5 min.

3.4. Identification of quercetin by FTIR

The IR spectra of isolated compound (quercetin) from polyherbal tablet were displayed in Fig. 3. The spectrum was typical of a flavonoid structure. The spectra were recorded in the range from 4000 to 600 cm⁻¹ and showed identical absorption bands with maxima at the following frequencies (max): 3392.35, 2380.76, 1658.94, 1446.45, 1378.65 cm⁻¹. The peaks at 3392 and 2919 cm⁻¹ are characteristic of the aromatic ring structure. The absorption maxima at 1610 and 1557 cm⁻¹ indicate the presence of a quinoid structure and -C=C- bonds, respectively. The absorption maximum at 1375 cm⁻¹ is due to phenolic OH groups. Melting point of isolated compound was found to be 314 °C.

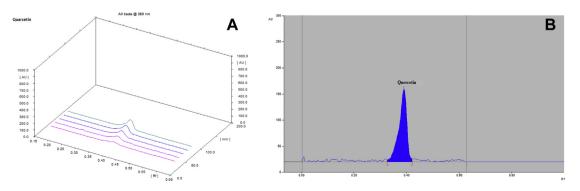


Fig. 1. 3D-chromatogram (A) and densitogram (B) of standard quercetin (0.5-2.5 µg/spot).

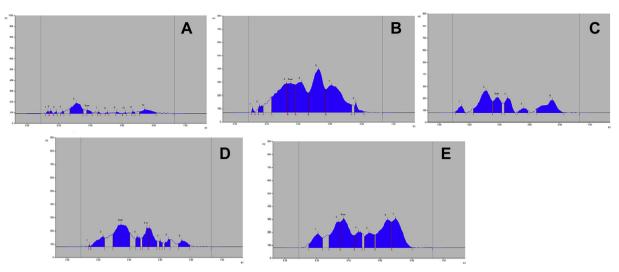


Fig. 2. HPTLC densitogram of acetone fraction of *Butea monosperma* (A), ethyl acetate fraction of *Bauhinia variegata* (B), n-butanol fraction of *Bauhinia variegata* (C), dichloromethane fraction of *Ocimum gratissimum* (D), ethyl acetate fraction of *Ocimum gratissimum* (E) showing quercetin.

Table 2
Pre- and post-compression evaluation of polyherbal tablet formulations PF1, PF2.

SN	Quality control tests	PF1	PF2
Pre-compres	ssion parameters of evaluation of polyherbal tablet formulations PF1,	PF2	
1.	Angle of repose (mean \pm S.D.) ($n = 3$)	29.3 ± 1.53	28.7 ± 0.58
2.	Bulk density (mean \pm S.D.) ($n = 3$) (g/cc)	0.427 ± 0.003	0.450 ± 0.003
3.	Tapped density (mean \pm S.D.) ($n = 3$) (g/cc)	0.515 ± 0.005	0.523 ± 0.008
4.	Compressibility index (%) (mean \pm S.D.) ($n = 3$)	17.15 ± 0.2	15.31 ± 0.12
Post-compre	ession evaluation of tablets		
5.	Thickness (mm) (mean \pm S.D.) ($n = 10$)	5.5 ± 0.02	5.6 ± 0.03
6.	Diameter (mm) (mean \pm S.D.) ($n = 10$)	11 ± 0.04	11 ± 0.05
7.	Hardness (kg/cm ²) (mean \pm S.D.) ($n = 10$)	3.1 ± 0.19	$\textbf{3.3} \pm \textbf{0.12}$
8.	Disintegration time (min) (mean \pm S.D.) ($n = 6$)	2.10 ± 0.42	$\textbf{4.25} \pm \textbf{1.12}$
9.	Average weight (mg) (mean \pm S.D.) ($n = 10$)	506 ± 3.77	500.6 ± 0.02
10.	Friability (%) (mean \pm S.D.) ($n = 3$)	0.3 ± 0.23	0.2 ± 0.16
Physical par	ameters (appearance) of tablet formulations		
11.	Color	Light greenish with few white spots	Light greenish with slightly
		slightly dark greenish spots	dark greenish spots
12.	Odor	Characteristic	Characteristic
13.	Shape	Circular biconvex tablet	Circular biconvex tablet
14.	Taste	Characteristic	Characteristic

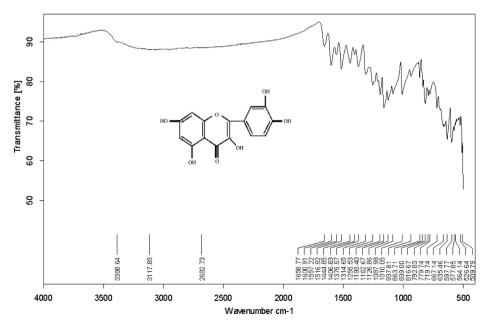


Fig. 3. IR spectrum of isolated compound from the mixture of fractions of Butea monosperma, Bauhinia variegata and Ocimum gratissimum.

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Table 3	
Effect of polyherbal formulation PTF-1 and PTF-2 on biochemical parameters in paracetamol in	duced hepatic injury in rats.

Treatment	SGPT U/L	SGOT U/L	ALP U/L	Total bilirubin (mg/dl)	Direct bilirubin (mg/dl)
Control	69.75 ± 1.96	135.67 ± 1.63	145.47 ± 3.73	0.89 ± 0.07	0.23 ± 0.01
Paracetamol treated	$284.25 \pm 1.38^{**}$	$421.18 \pm 1.12^{***}$	$464.43\pm1.36^{***}$	$3.74 \pm 0.92^{***}$	$0.65 \pm 0.03^{***}$
Silymarin treated	$71.46 \pm 0.81^{\#\#}$	$142.63 \pm 0.82^{\#\#}$	$174.73\pm1.24^{\#\#}$	$1.25 \pm 0.83^{\#}$	$0.29 \pm 0.05^{\#\#}$
PTF-1 treated	$75.85 \pm 1.35^{\#\#}$	$149.63\pm1.73^{\#\#}$	$180.64 \pm 1.25^{\text{NS}}$	$1.48\pm0.42^{\text{NS}}$	$0.32\pm0.03^{\text{NS}}$
PTF-2 treated	$65.56 \pm 1.68^{\#\#}$	$152.98 \pm 1.48^{\#\#}$	$167.87 \pm 1.78^{\#\#}$	$1.32\pm0.78^{\#\#}$	$0.33 \pm 0.10^{\#\#}$

В

4.5

3.5

2.5

1.5

3

2

1

0

Control

PCM

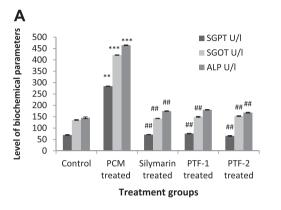
treated

0.5

4

Level of biochemical parameters

***P < 0.001 vs control, ${}^{\#}P < 0.05$, ${}^{\#\#}P < 0.01$ vs PCM treated, NS non significant.



Effect of PTF-1 and PTF-2 on SGOT, SGPT and ALP(U/L) in paracetamol induced hepatic injury in rats.

Effect of PTF-1 and PTF-2 on SGOT Total Bilirubin (mg/dl) in paracetamol induced hepatic injury in rats.

Silymarin

treated

Treatment groups

Fig. 4. Graphs showing effect of polyherbal formulation PTF-1 and PTF-2 on biochemical parameters in paracetamol induced hepatic injury in rats.

3.5. Evaluation of hepatoprotective activity

From the results of acute toxicity study by previously published literature, 2000 mg/kg was considered as maximum tolerable dose (MTD) of alcoholic extract of *B. monosperma*,¹⁴ *B. variegata*²⁶ and *O. gratissimum*.²⁷ 1/20th of this dose was taken as experimental dose

for subsequent hepatoprotective studies. The effective dose of each fraction was decided on the basis of previously published report, i.e. 100 mg/kg of body weight.

Total bilirubin

PTF-1

±±

PTF-2

(mg/dl)

Paracetamol has enhanced the levels of SGPT, SGOT, bilirubin (both total and direct bilirubin levels), alkaline phosphatase level (ALP). The results indicated that the polyherbal tablet formulations

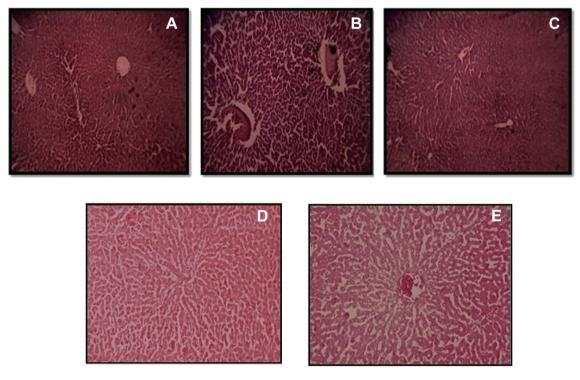


Fig. 5. The photomicrographs of liver section from rats (A) received saline 0.5 mL as normal control group (10×10), (B) received paracetamol (2 g/kg) (10×10), (C) received silymarin (100 mg/kg bw) + PCM (10×10), (D) received PF1 (50 mg/kg bw) + PCM and (E) received PF2 (100 mg/kg bw).

Table 4

Histopathological changes in the liver of rats treated with polyherbal formulation PTF-1 and PTF-2.

Microscopic observation	Control	PCM	Silymarin	PTF-1	PTF-2
Degeneration in hepatocytes	+	+++	++	++	++
Hepatic cell injury	Ø	+++	++	+	++
Focal necrosis	Ø	+++	Ø	$+ \perp$	+
Congestion in central vein	+	+++	+	$+ \perp$	$+ \perp$
Vascular swelling	+	+++	+	+	$+ \perp$
Kupffer cell proliferation	Ø	++	+	$^{++}$	+

Ø, absent; \perp , few; +, mild; ++, moderate; +++, severe; ++++, extremely severe, PCM – paracetamol.

(PTF-1 and PTF-2) significantly reduced the elevated levels of SGPT, SGOT, ALP and bilirubin when compared to paracetamol treated group. The formulation PTF-1 has reduced the increased SGPT level from 284.25 U/L to 75.85 U/L, SGOT level from 421.18 U/L to 149.63 U/L, ALP level from 464.43 U/L to 180.64 U/L and bilirubin level from 3.74 mg/dl to 1.48 mg/dl. The formulation PTF-2 has reduced the increased SGPT level from 284.25 U/L to 65.56 U/L, SGOT level from 421.18 U/L to 152.98 U/L, ALP level from 464.43 U/L to 167.87 U/L and bilirubin level from 3.74 mg/dl.

Hepatic cells participate in metabolic activities and contain host of enzymes. In tissue, SGOT and SGPT exist in mitochondria. In liver injury, transport function of the hepatocytes gets disturbed, resulting in the leakage of plasma membrane and thereby causing an increased enzyme level in serum. The elevated activities of these enzymes are indicative of cellular leakage and the functional integrity of the cell membranes in liver. ALP is excreted by liver via bile in the liver injury due to hepatotoxins, which results in a defective excretion of bile by the liver and is reflected in their increased levels in serum. In drug-induced liver toxicity, the level of total and direct bilirubin gets elevated. Both formulations showed marked decrease in the elevated levels of SGOT, SGPT, ALP and bilirubin (total and direct) which is nearer to the levels of control group.²⁸ The results are shown in Table 3, Fig. 4.

Histopathological examination of the liver sections confirmed that the normal liver (Fig. 5A) architecture was damaged with paracetamol administration (Table 4, Fig. 5B). However, pretreatment of the PTF1 (50 mg/kg) and PTF2 (100 mg/kg) dose significantly reduced the severity of histopathological injury (compared with the paracetamol group). The results of the biochemical tests and histopathological observations suggest that PTF2 formulation is better than PTF1.

Rats treated with silymarin, PTF-1 and PTF-2 formulations showed noticeable improvement in histopathological parameters (Table 4 and Fig. 5C–E). Therefore, both the formulations have

Table 5

Stability studies data for PTF-2 formulation at 25 °C/60% RH.

Parameter	PTF-2				
	0 day	10 days	20 days	30 days	
Stability studies data for PTF-2 formulation at 25 °C/60% RH					
Appearance	*	**	**	**	
Hardness (kg/cm ²)	$\textbf{3.3} \pm \textbf{0.12}$	$\textbf{3.3} \pm \textbf{0.15}$	$\textbf{3.3} \pm \textbf{0.12}$	$\textbf{3.6} \pm \textbf{0.06}$	
(n = 3) (mean ± S.D.)					
Disintegration time	$\textbf{2.12} \pm \textbf{0.12}$	$\textbf{3.32} \pm \textbf{0.15}$	$\textbf{2.33} \pm \textbf{0.12}$	$\textbf{2.20} \pm \textbf{0.06}$	
$(\min)(n = 3)$					
$(mean \pm S.D.)$					
Stability studies data for F	TF-2 formulati	ion 40 °C/75%	RH		
Appearance	*	**	**	***	
Hardness (kg/cm ²)	$\textbf{3.4} \pm \textbf{0.12}$	$\textbf{3.3} \pm \textbf{0.15}$	$\textbf{3.5} \pm \textbf{0.23}$	$\textbf{3.8} \pm \textbf{0.15}$	
(n = 3)					
Disintegration time	4.00 ± 0.58	$\textbf{4.30} \pm \textbf{0.83}$	$\textbf{4.36} \pm \textbf{0.86}$	5.30 ± 1.69	
$(\min)(n = 3)$					

*Light greenish color with slightly dark greenish spots, **No change, ***Light greenish color with dark greenish spots.

prominent action on paracetamol-induced liver damage. Moreover, at necropsy, livers of rats treated with paracetamol appeared degeneration in hepatocytes, hepatic cell injury, focal necrosis, congestion in central vein, vascular swelling, and Kupffer cell proliferation. Furthermore, no gross pathological findings were noted in the livers of the other groups of rats.

3.6. Stability studies

Selected polyherbal tablet formulation PTF-2 was found to be stable when subjected to stability studies at 25 °C/60% RH and 40 °C/75% RH for a time period of 30 days. There was no significant change in the physicochemical properties of tablets i.e. appearance, disintegration time and hardness of the tablets. The results are shown in Table 5.

4. Conclusion

Biochemical marker showed improved results for polyherbal tablet formulation at 100 mg/kg. Polyherbal tablet formulation contains a potent hepatoprotective agent suggested to be a flavone concentrated in polyherbal formulation which may find clinical application in amelioration of paracetamol induced liver damage.

Authors contribution

Principle investigator and project leader: Mrs. Arti Gupta. Concept development: Dr. Navin R. Sheth. Analysis and written the manuscript: Mrs. Sonia Pandey. Statistical analysis: Mr. Jitendra S. Yadav and Dr. Dinesh Shah.

Conflicts of interest

All authors have none to declare.

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